



Supplementary Materials

Exploring the Anticancer Potential of *Premna resinosa* (Hochst.) Leaf Surface Extract: Discovering New Diterpenes as Heat Shock Protein 70 (Hsp70) Binding Agents

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Citation: Parisi, V.; Donadio, G.; Bellone, M.L.; Belaabed, S.; Bader, A.; Bisio, A.; Iobbi, V.; Gazzillo, E.; Chini, M.G.; Bifulco, G.; et al. Exploring the Anticancer Potential of *Premna resinosa* (Hochst.) Leaf Surface Extract: Discovering New Diterpenes as Heat Shock Protein 70 (Hsp70) Binding Agents. *Plants* **2023**, *12*, 2421. https://doi.org/10.3390/ 10.3390/plants12132421

Academic Editor: Maria José U. Ferreira

Received: 16 May 2023 Revised: 12 June 2023 Accepted: 20 June 2023 Published: 22 June 2023



Copyright: © 2023 by the authors. Submitted for possible open access publication under the terms and conditions of the Creative Commons Attribution (CC BY) license (https://creativecommons.org/license s/by/4.0/). **Figure S1.** ¹H NMR spectrum of compound **1** (CD₃OD, 600 MHz).

Figure S2. ¹³C NMR spectrum of compound 1 (CD₃OD, 600 MHz).

Figure S3. COSY spectrum of compound 1 (CD₃OD, 600 MHz).

Figure S4. HSQC spectrum of compound 1 (CD₃OD, 600 MHz).

Figure S5. HMBC spectrum of compound 1 (CD₃OD, 600 MHz).

Figure S6. NOESY spectrum of compound **1** (CD₃OD, 600 MHz)**Figure S7.** HRESIMS of compound **1**.

Figure S8. ¹H NMR spectrum of compound 2 (CD₃OD, 600 MHz).

Figure S9. ¹³C NMR spectrum of compound **2** (CD₃OD, 600 MHz).

Figure S10. COSY spectrum of compound 2 (CD₃OD, 600 MHz).

Figure S11. HSQC spectrum of compound 2 (CD₃OD, 600 MHz).

Figure S12. HMBC spectrum of compound 2 (CD₃OD, 600 MHz).

Figure S13. HRESIMS of compound 2.

Figure S14. 2D structures of investigated stereoisomers of 1 and 4.

Figure S15. Binding pose and interaction of 3 docked to Hsp70 ATP binding site.

Figure S16. Molecular dynamic simulation results.

Table S1. ¹H experimental and calculated NMR chemical shifts for **1a-b**, with $a|\Delta\delta|({}^{1}\text{H})$ and ${}^{c}\text{MAE}$ values. Chemical shift data here reported were produced using benzene as reference compound for sp² hydrogens, and tetramethylsilane (TMS) for sp³ hydrogens.

Table S2. ¹³C experimental and calculated NMR chemical shifts for **1a-b**, with $a|\Delta\delta|(^{13}C)$ and ^{b}MAE values. Chemical shift data here reported were produced using benzene as reference compound for sp² carbons, and tetramethylsilane (TMS) for sp³ carbons.

Table S3. ¹H experimental and calculated NMR chemical shifts for **4a-d**, with $a|\Delta\delta|({}^{1}\text{H})$ and ${}^{c}\text{MAE}$ values. Chemical shift data here reported were produced using benzene as reference compound for sp² hydrogens, and tetramethylsilane (TMS) for sp³ hydrogens.

Table S4. ¹³C experimental and calculated NMR chemical shifts for **4a-d**, with $|\Delta \delta|(^{13}C)$ and ^{b}MAE values. Chemical shift data here reported were produced using benzene as reference compound for sp² carbons, and tetramethylsilane (TMS) for sp³ carbons.



Figure S2. ¹³C NMR spectrum of compound 1 (CD₃OD, 600 MHz).



Figure S3. COSY spectrum of compound 1 (CD₃OD, 600 MHz).



Figure S4. HSQC spectrum of compound 1 (CD₃OD, 600 MHz).



Figure S5. HMBC spectrum of compound 1 (CD₃OD, 600 MHz).



Figure S6. NOESY spectrum of compound 1 (CD₃OD, 600 MHz)



Figure S7. HRESIMS of compound 1.



Figure S8. ¹H NMR spectrum of compound 2 (CD₃OD, 600 MHz).



Figure S9. ¹³C NMR spectrum of compound 2 (CD₃OD, 600 MHz).



Figure S10. COSY spectrum of compound 2 (CD₃OD, 600 MHz).





Figure S11. HSQC spectrum of compound 2 (CD₃OD, 600 MHz).



Figure S12. HMBC spectrum of compound 2 (CD₃OD, 600 MHz).

-50



Figure S13. HRESIMS of compound 2.

,ОН



Figure S14. 2D structures of investigated stereoisomers of 1 and 4.



Figure S15. Binding pose and interaction of **3** docked to Hsp70 ATP binding site. (**A**) The protein is reported as grey ribbons and residues are colored by atom types; the ligand is reported as green capped sticks; H-bonds are presented as yellow dotted lines. (**B**) The ligand is surrounded by the protein residues represented as follows: the negatively charged residues are indicated in red, polar residues are in cyan, hydrophobic residues are shown in green; H-bonds are depicted as purple arrows.



Figure S16. Molecular dynamic simulation results. (**A**) Root-mean square deviation (RMSD) plot for **3**/Hsp70 complex along 100 ns molecular dynamics simulation related to $C\alpha$ positions of residues belonging to the protein backbone (blue) and the ligand (purple). (**B**) Protein-ligand interactions (or 'contacts') plot for **3**/Hsp70 complex along 100 ns molecular dynamics simulation. Contacts are categorized into four types: hydrogen bonds, hydrophobic, ionic and water bridges. (**C**) Ligand atom interactions with the protein residues. Interactions that occur more than 30.0% of the simulation time in the selected trajectory (0.00 through 100.00 ns), are shown.

δ _{exp} (1H), ppm	$\delta_{calc}(^{1}H)$	I), ppm	Δδ (1H), ppm			
exp_1	calc_1a calc_2b		calc_1a	calc_1b		
1.74	1.00	1.64	0.74	0.10		
1.01	0.87	0.99	0.14	0.02		
1.46	1.46 1.68 1.69		0.22	0.23		
1.5	1.22	1.32	0.28	0.18		
1.46	1.19	1.24	0.27	0.22		
1.24	1.32	1.36	0.08	0.12		
1.33	1.38	1.38	0.05	0.05		
1.96	2.07	2.03	0.11	0.07		
2.05	2.07	2.13	0.02	0.08		
5.43	5.37	5.75	0.06	0.02		
2.71	2.56	2.81	0.15	0.10		
2.94	3.32	2.95	0.38	0.01		
2.63	2.31	2.61	0.32	0.02		
3.3	3.17	3.22	0.13	0.08		
5.94	6.14	6.46	0.20	0.52		
6.3	6.43	6.75	0.13	0.11		
1.46	1.61	1.34	0.15	0.12		
0.93	0.97	0.98	0.04	0.05		
	 δexp(1H), ppm exp_1 1.74 1.01 1.46 1.5 1.46 1.24 1.33 1.96 2.05 5.43 2.71 2.94 2.63 3.3 5.94 6.3 1.46 0.93 	ðexp(¹H) ðeale(¹H) ppm ðeale(¹H) exp_1 calc_1a 1.74 1.00 1.74 1.00 1.01 0.87 1.46 1.68 1.5 1.22 1.46 1.19 1.24 1.32 1.33 1.38 1.96 2.07 2.05 2.07 5.43 5.37 2.71 2.56 2.94 3.32 2.63 2.31 3.3 3.17 5.94 6.14 6.3 6.43 1.46 1.61	besp(III) besp(III) ppm 6calc(III), ppm exp_1 calc_1a 1.74 1.00 1.74 0.87 1.01 0.87 1.46 1.68 1.46 1.68 1.46 1.22 1.46 1.19 1.24 1.32 1.46 1.32 1.46 1.19 1.24 1.32 1.46 1.19 1.24 1.32 1.46 1.32 1.24 1.32 1.33 1.38 1.96 2.07 2.05 2.07 2.05 2.07 2.71 2.56 2.71 2.56 2.63 2.31 2.63 2.31 2.63 3.17 3.3 3.17 3.4 6.46 6.3 6.43 6.3 6.43 6.43 6.75	$\delta_{exp}(1H)$ $\beta_{eabc}(1H)$, ppm $ \Delta\delta (1H)$ ppm $Calc_1a$ $calc_2b$ $calc_1a$ exp_1 $calc_1a$ $calc_2b$ $calc_1a$ 1.74 1.00 1.64 0.74 1.01 0.87 0.999 0.14 1.46 1.68 1.69 0.22 1.46 1.68 1.69 0.22 1.46 1.68 1.69 0.22 1.46 1.68 1.69 0.22 1.46 1.68 1.69 0.22 1.46 1.68 1.69 0.22 1.46 1.19 1.24 0.27 1.24 1.32 1.38 0.05 1.33 1.38 1.38 0.05 1.96 2.07 2.03 0.11 2.05 2.07 2.03 0.15 2.05 2.07 2.03 0.15 2.05 2.07 2.03 0.13 2.05 2.31 2.61		

0.88

0.98

19

20

MAE

0.96

0.87

0.87

1.09

0.09

0.22

0.19

0.08

0.11

0.11

Table S1. ¹H experimental and calculated NMR chemical shifts for 1**a-b**, with ^a $|\Delta\delta|(^{1}H)$ and ^cMAE values. Chemical shift data here reported were produced using benzene as reference compound for sp² hydrogens, and tetramethylsilane (TMS) for sp³ hydrogens.

¹³ C	δ _{exp} (¹³ C),	δ _{calc} (13C	C), ppm	Δδ (¹³ C), ppm		
Position	exp_1	calc_1a calc_1b		calc_1a	calc_1b	
1	38.7	37.6	39.8	1.1	1.1	
2	18.0	20.8	20.9	2.8	2.9	
3	41.7	42.1	41.7	0.4	0.0	
4	33.0	33.8	34.2	0.8	1.2	
5	50.1	42.1	49.6	8.0	0.5	
6	23.5	26.5	26.2	3.0	2.7	
7	122.0	126.3	126.3	4.3	4.3	
8	134.0	138.9	138.9	4.9	4.9	
9	48.6	47.1	49.0	1.5	0.4	
10	36.0	37.8	37.5	1.8	1.5	
11	35.0	39.7	36.9	4.7	1.9	
12	203.0	201.3	201.3	1.7	1.7	
13	144.0	144.8	144.8	0.8	0.8	
14	37.2	39.5	38.6	2.3	1.4	
15	174.5	171.2	171.2	3.3	3.3	
16	125.5	134.5	134.5	9.0	9.0	
17	21.1	24.1	23.8	3.0	2.7	
18	32.0	33.6	33.9	1.6	1.9	
19	21.0	23.1	22.9	2.1	1.9	
20	13.0	22.8	16.7	9.8	3.7	
MAE				3.3	2.4	

Table S2. ¹³C experimental and calculated NMR chemical shifts for **1a-b**, with $a|\Delta\delta|(^{13}C)$ and ^{b}MAE values. Chemical shift data here reported were produced using benzene as reference compound for sp² carbons, and tetramethylsilane (TMS) for sp³ carbons.

${}^{1}\mathrm{H}$	δ _{exp} (1H),	δ _{cale} (1H), ppm			Δδ (¹ H), ppm				
	ppm								
Position	exp_4	calc_4a	calc_4b	calc_4c	calc_4d	calc_4a	calc_4b	calc_4c	calc_4d
1	0.87	1.02	1.11	1.71	1.54	0.15	0.24	0.84	0.67
1	1.76	1.75	1.71	1.33	1.17	0.01	0.05	0.43	0.59
2	1.42	1.36	1.36	1.43	1.35	0.06	0.06	0.01	0.07
2	1.6	1.72	1.71	1.81	1.78	0.12	0.11	0.21	0.18
3	1.45	1.37	1.37	1.44	1.37	0.08	0.08	0.01	0.08
3	1.21	1.24	1.25	1.35	1.27	0.03	0.04	0.14	0.06
5	1.25	1.36	1.38	1.65	1.64	0.11	0.13	0.40	0.39
6	1.91	2.03	2.05	2.22	2.15	0.12	0.14	0.31	0.24
6	2.02	2.15	2.13	2.16	2.10	0.13	0.11	0.14	0.08
7	5.48	5.46	5.49	5.67	5.53	0.02	0.01	0.19	0.05
9	1.87	2.38	2.28	1.72	1.51	0.51	0.41	0.15	0.36
11	1.71	1.61	1.56	2.01	1.96	0.16	0.15	0.24	0.19
11	1.77	1.46	1.82	1.83	1.60	0.25	0.05	0.12	0.11
12	4.71	4.60	4.44	5.44	4.91	0.11	0.27	0.73	0.20
14	6.05	5.72	5.70	5.68	5.72	0.33	0.35	0.37	0.33
16	5.01	4.89	4.90	4.91	4.91	0.12	0.11	0.10	0.10
16	5.03	4.86	4.93	4.90	4.95	0.17	0.10	0.13	0.08
17	1.78	1.70	1.66	1.87	1.79	0.08	0.12	0.09	0.01
18	0.89	0.86	0.87	0.91	0.87	0.03	0.02	0.02	0.02
19	0.93	0.97	0.99	1.01	0.98	0.04	0.06	0.08	0.05
20	0.83	0.93	0.97	1.04	1.04	0.10	0.14	0.21	0.21
MAE						0.13	0.13	0.23	0.19

Table S3. ¹H experimental and calculated NMR chemical shifts for **4a-d**, with ^a $|\Delta\delta|$ (¹H) and ^cMAE values. Chemical shift data here reported were produced using benzene as reference compound for sp² hydrogens, and tetramethylsilane (TMS) for sp³ hydrogens.

¹³ C	δ _{exp} (¹³ C), ppm	δ _{calc} (¹³ C), ppm			Δδ (¹³ C), ppm				
Position	exp_4	calc_4a	calc_4b	calc_4c	calc_4d	calc_4a	calc_4b	calc_4c	calc_4d
1	39.0	39.3	39.5	37.5	36.0	0.3	0.5	1.5	3.0
2	18.4	20.8	20.8	20.9	20.9	2.4	2.4	2.5	2.5
3	42.0	41.6	41.6	42.3	42.2	0.4	0.4	0.3	0.2
4	36.4	34.2	34.0	34.5	34.1	2.2	2.4	1.9	2.3
5	50.1	49.5	49.8	45.2	42.0	0.6	0.3	4.9	8.1
6	23.4	26.3	26.4	26.7	26.9	2.9	3.0	3.3	3.5
7	122.3	127.5	127.4	130.2	129.0	5.2	5.1	7.9	6.7
8	134.8	138.4	138.3	140.8	138.9	3.6	3.5	6.0	4.1
9	50.3	49.6	52.3	51.0	51.5	0.7	2.0	0.7	1.2
10	36.0	37.9	38.9	38.2	38.7	1.9	2.9	2.2	2.7
11	33.3	35.2	32.8	36.5	40.7	1.9	0.5	3.2	7.4
12	69.1	70.0	71.3	69.9	69.7	0.9	2.2	0.8	0.6
13	176.0	179.9	178.8	179.8	179.5	3.9	2.8	3.8	3.5
14	114.5	117.0	117.3	115.8	116.5	2.5	2.8	1.3	2.0
15	176.5	172.5	172.6	172.9	172.8	4.0	3.9	3.6	3.7
16	71.5	70.3	70.5	69.9	70.1	1.2	1.0	1.6	1.4
17	21.4	23.6	24.5	25.2	25.7	2.2	3.1	3.8	4.3
18	32.0	33.8	33.9	33.9	33.3	1.8	1.9	1.9	1.3
19	21.0	23.1	22.9	23.6	23.0	2.1	1.9	2.6	2.0
20	13.0	15.9	15.5	23.6	22.8	2.9	2.5	10.6	9.8
MAE						2.2	2.3	3.2	3.5

Table S4. ¹³C experimental and calculated NMR chemical shifts for **4a-d**, with ^a $|\Delta\delta|(^{13}C)$ and ^bMAE values. Chemical shift data here reported were produced using benzene as reference compound for sp² carbons, and tetramethylsilane (TMS) for sp³ carbons.